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Separation of Racemic and Meso-1,2-Bis(2-[(2-Bromo-4,5- Dimethoxyphenyl)Hydroxymethyl]-4,5- Dimethoxybenzyl]-4,5- Dimethoxybenzenes by Host-Guest Inclusion

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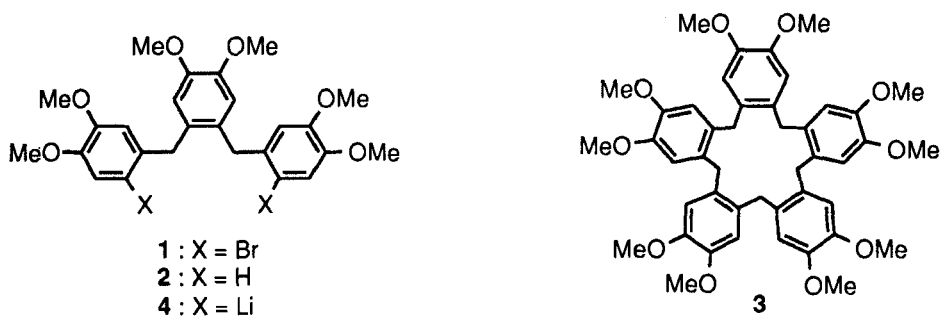
SEPARATION OF RACEMIC AND MESO-1,2-BIS{2-[(2-BROMO-4,5-DIMETHOXYPHENYL)HYDROXYMETHYL]-4,5-DIMETHOXYBENZYL}-4,5-DIMETHOXYBENZENES BY HOST-GUEST INCLUSION

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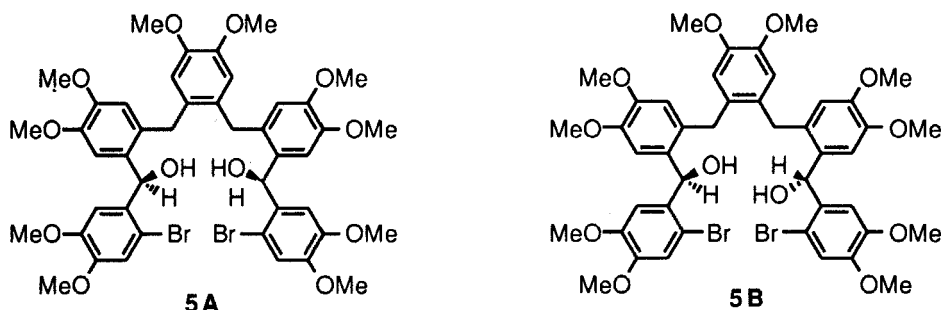
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Abstract Diastereoisomers of title compounds were separated by inclusion of solvent and their stereostructures were determined by $^1\text{H-NMR}$ measurement using a chiral shift reagent, $\text{Eu}(\text{tfc})_3$.

In the course of synthetic study on orthocyclophanes, we have adopted a divergent method in which reaction of veratrol with 6-bromoveratryl alcohol in the presence of acid gives a core compound (1),^{1, 2} and then 6-bromoveratraldehyde (2)² serves as a key building block for extension of benzyl unit.



In one step for synthesis of cyclopentaveratrylene (**3**),² a dilithiolide (**4**), derived from **1** and *n*-BuLi, was reacted with 2.1 eq. of **2** at -50 °C for 2 h to give 1,2-bis{2-[(2-bromo-4,5-dimethoxyphenyl)hydroxymethyl]-4,5-dimethoxybenzyl}-4,5-dimethoxybenzene (**5**), mp 107-113 °C, in 73 % yield. Plain structure of the product was confirmed on the basis of IR spectrum [3585 cm⁻¹ (OH)] and mass spectrum [*m/z* 926 (M⁺), 928 (M⁺+2), 930 (M⁺+4); 1 : 2 : 1 ratio]. Although the product showed one spot on thin layer chromatography under several conditions, its ¹H-NMR spectrum revealed it was a mixture of diastereoisomers (1 : 1 ratio) (Chart 1). Synthesis of the target compound (**3**) was accomplished in three steps from **5** without separation.



When purification of **5** was carried out, a novel character of the dibromodiol structure was observed. Recrystallization of the mixture from AcOEt-hexane furnished one isomer (**5A**)³ as 1 : 1 adduct with AcOEt. On the other hand, the residue was recrystallized from Et₂O-acetone to afford another isomer (**5B**)³ as 1 : 1 complex with Et₂O. Host-guest ratios of the complexes were estimated by integral ratio of aromatic protons of the hosts and methyl protons of the solvents on those ¹H-NMR spectra (Charts 2 and 3), respectively. Attempts to determine stereostructures of **5A** and **5B** by HPLC analysis using various types of chiral columns were unsuccessful. After all, we could clarify their stereostructures by 270 MHz ¹H-NMR measurement using a chiral shift reagent, Eu(tfc)₃ (Charts 4 and 5). Methine protons of benzhydryl moieties in **5B** were splitted by formation of diastereomers and those in **5A** not. Therefore, stereostructure of **5A** was decided to be *meso* and **5B** *racemic*, respectively.

Analogous compounds (**6**² and **7**⁴) have no ability of solvent inclusion. Therefore,

Chart 1
5 (mixture)
[100 MHz]

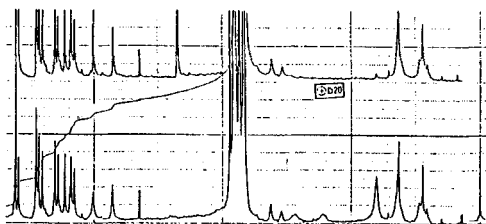


Chart 2
5A • AcOEt
[100 MHz]

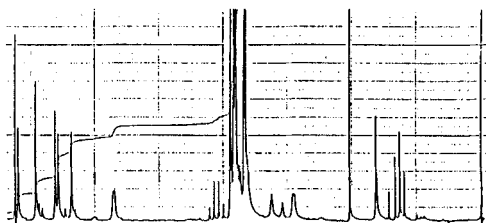


Chart 3
5B • Et₂O
[100 MHz]

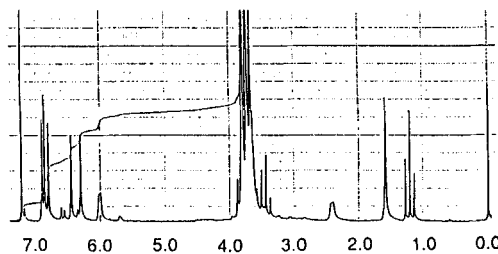


Chart 4
5A • AcOEt
+ Eu(tfc)₃
[270 MHz]

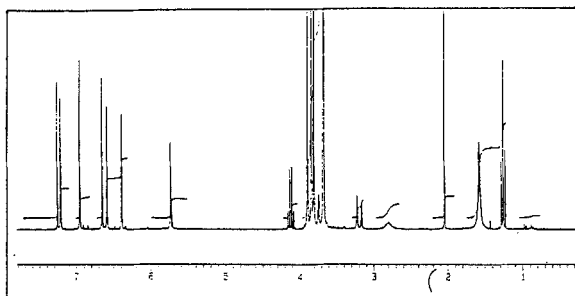
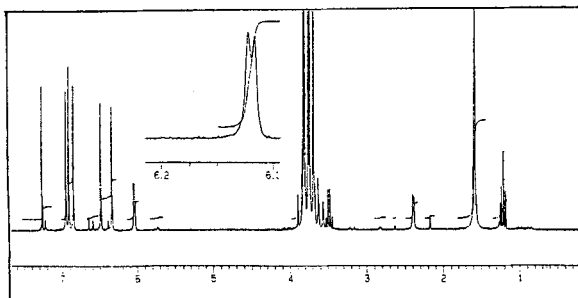
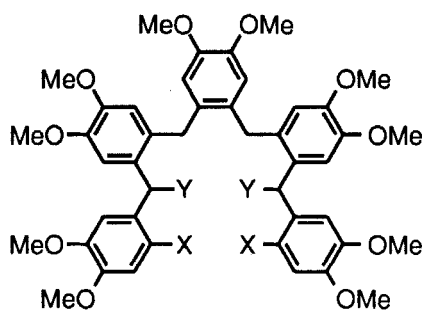


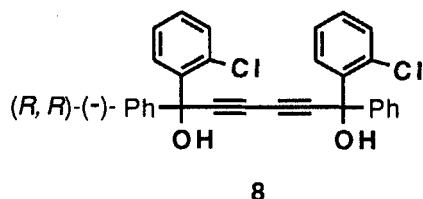
Chart 5
5B • Et₂O
+ Eu(tfc)₃
[270 MHz]



existence of both substituents (hydroxyl and bromo groups) are requisite in the inclusion of AcOEt or Et₂O. Other solvents (MeOH, benzene, hexane) employed for recrystallization of **5** were not included.



6 : X = Br, Y = H
7 : X = H, Y = OH



Toda and his group have energetically investigated on host-guest chemistry in the solid state.⁵ They have reported that a chiral host compound (**8**) has quite strong ability of solvent inclusion. Considering the facts, it is noticeable that **5** and **8** have bis(*o*-halogenobenzhydrol) moiety as a common structure. In contrast with flexible structure of **5**, rigid diacetylene skeleton and two hydroxyl groups existing apart from each other in **8** may generate excellent effect for guest trapping.⁶

REFERENCES AND NOTES

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2. H. Hara, H. Nakamura, S. Watanabe, and O. Hoshino, *Tetrahedron Lett.*, **35**, 526, (1994).
3. **5A**·AcOEt, mp 121-122 °C; **5B**·Et₂O, mp 197-200 °C.
4. **7** (diastereomixture) was prepared from **4** and veratraldehyde in a manner similar to that noted for **5**. **7**: an amorphous mass (82 %); ¹H-NMR δ 2.32, 2.92, (each brs, OH), 5.51, 5.69 (each s, CHOH); IR (CHCl₃) 3575 cm⁻¹ (OH); MS (m/z) 770 (M⁺).
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